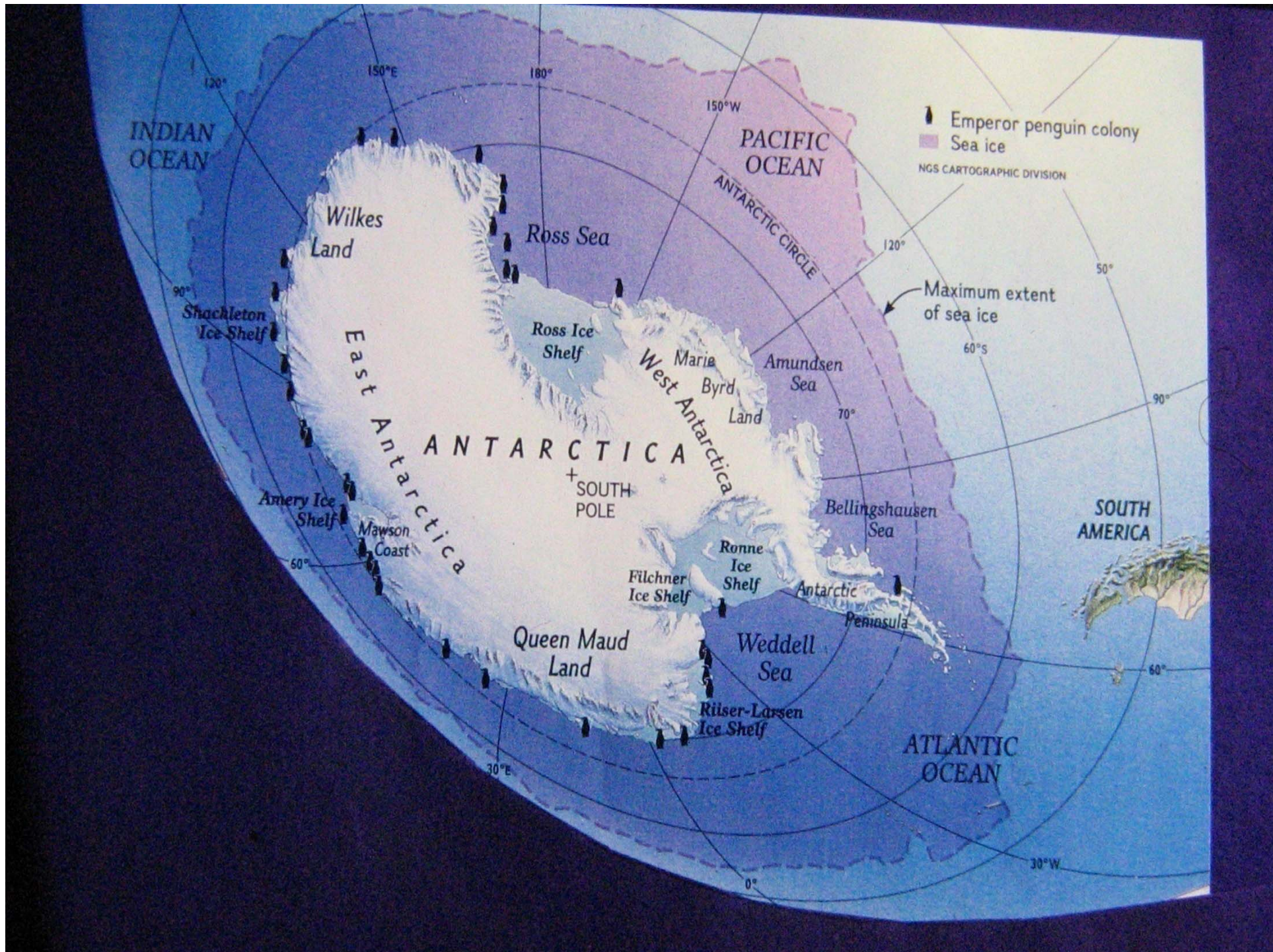




Water Quality : Lakes of Schirmacher Oasis, Antarctica

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Roorkee – 247 667 (Uttarakhand), India





Schirmacher Oasis, Antarctica



Maitree and Tirumala

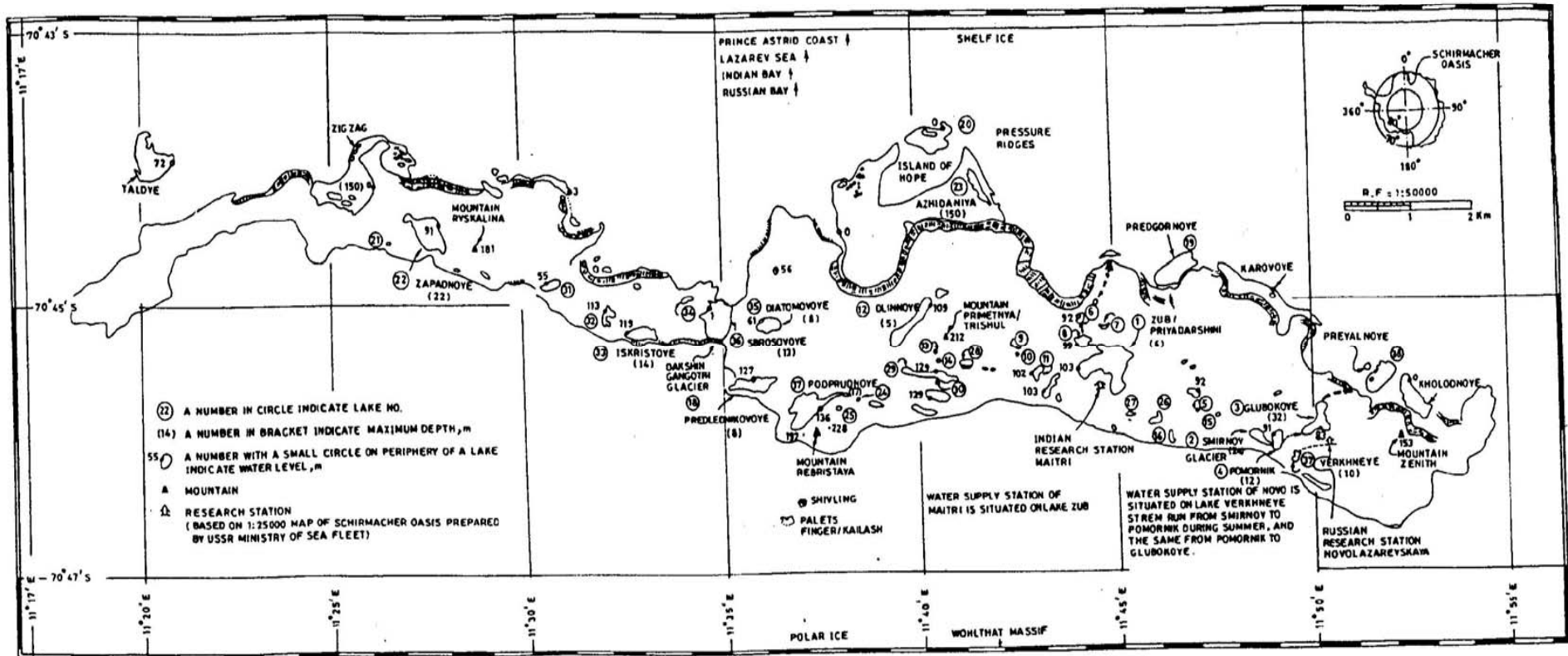


FIG. 1 STUDY AREA SHOWING THE LAKES SURVEYED IN SCHIRMACHER OASIS, ANTARCTICA

Classification of Lakes:

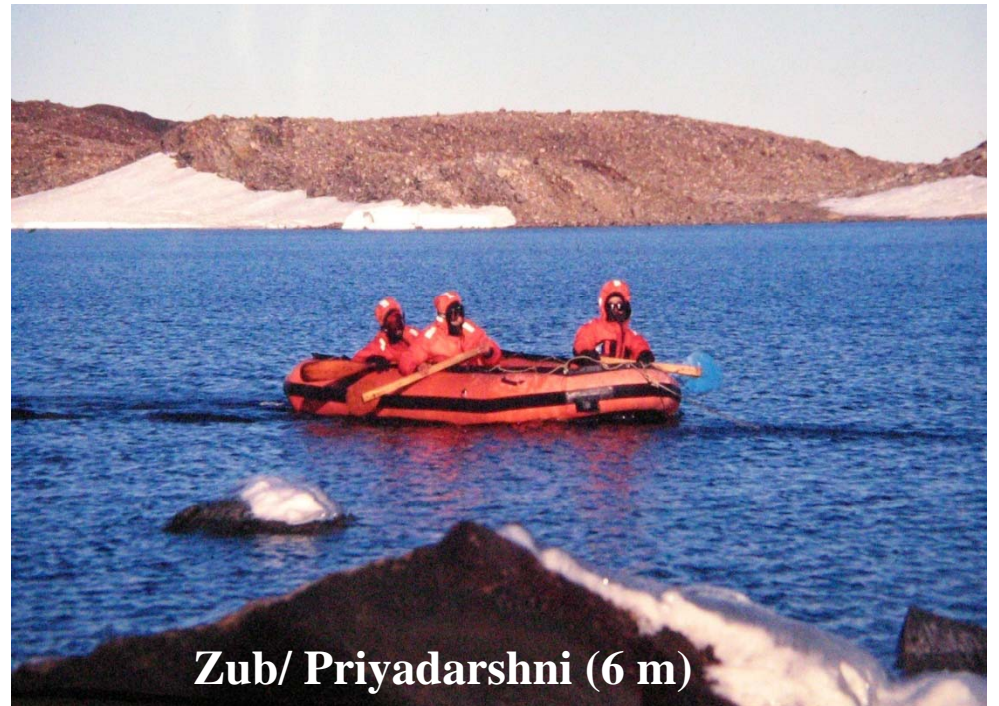
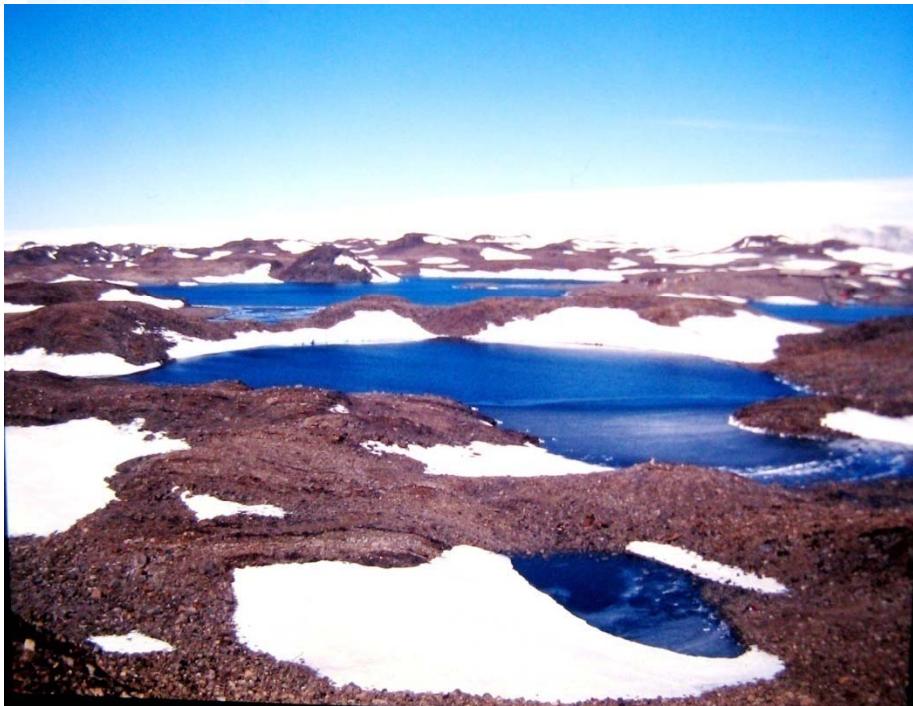
- Glacier-fed/Glacier-fed intermountainous
- Intermountainous
- Grounding line



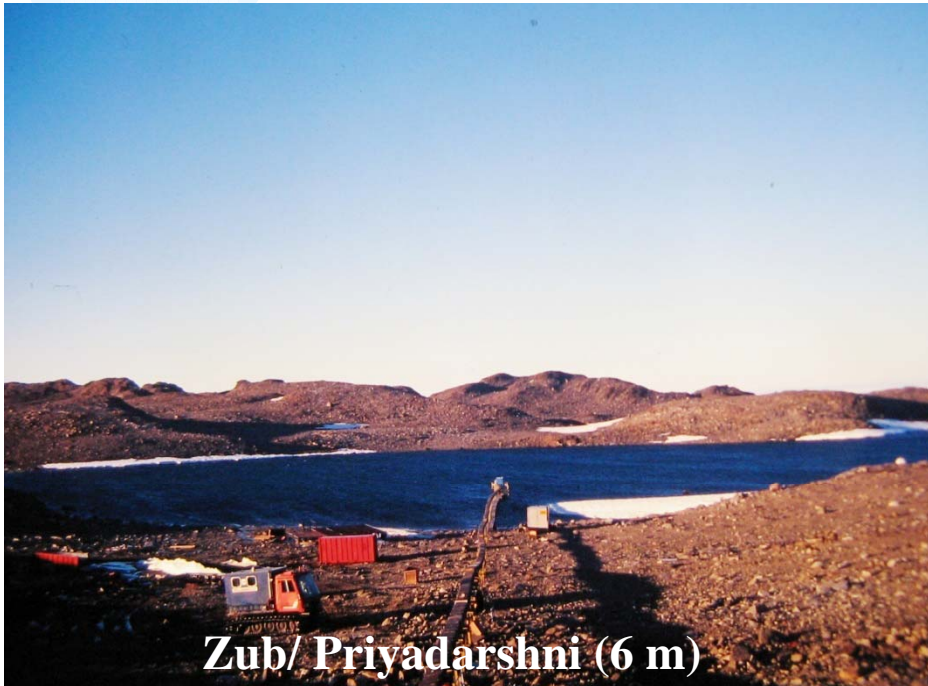
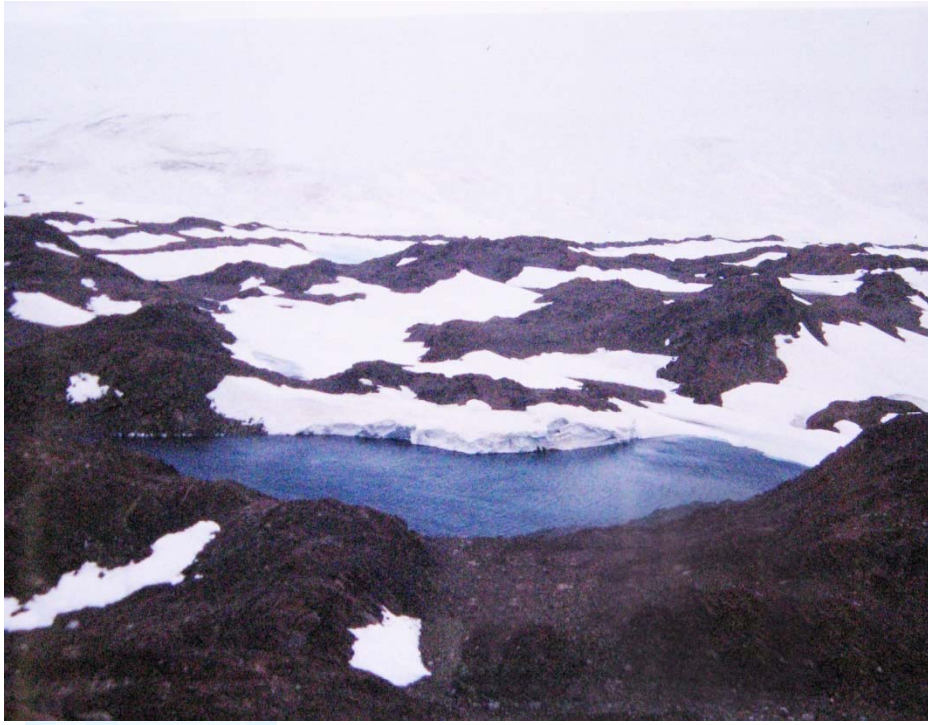




Glubokoye(36 m)



Zub/ Priyadarshni (6 m)



Zub/ Priyadarshni (6 m)

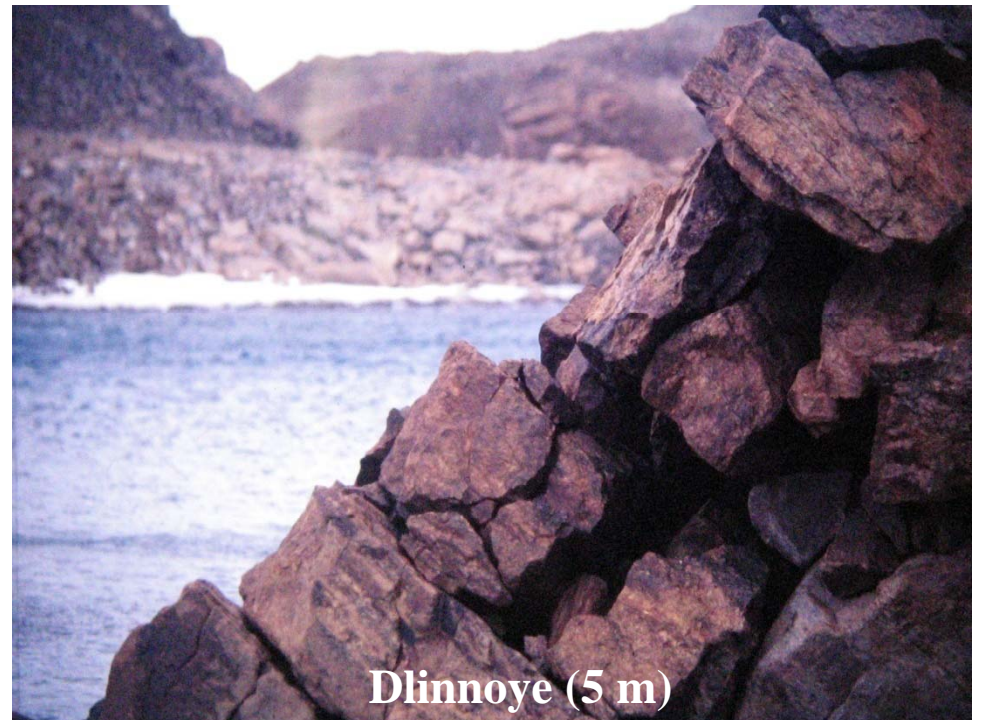




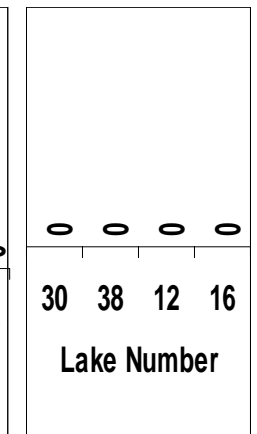
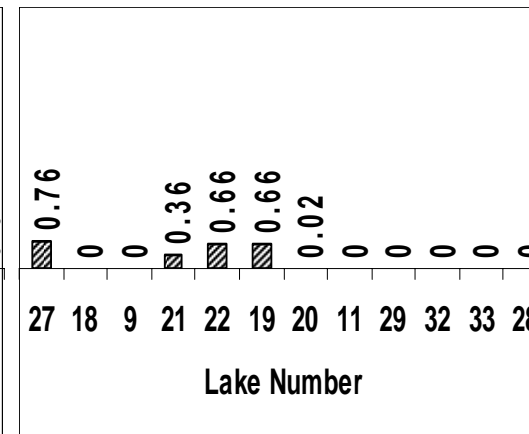
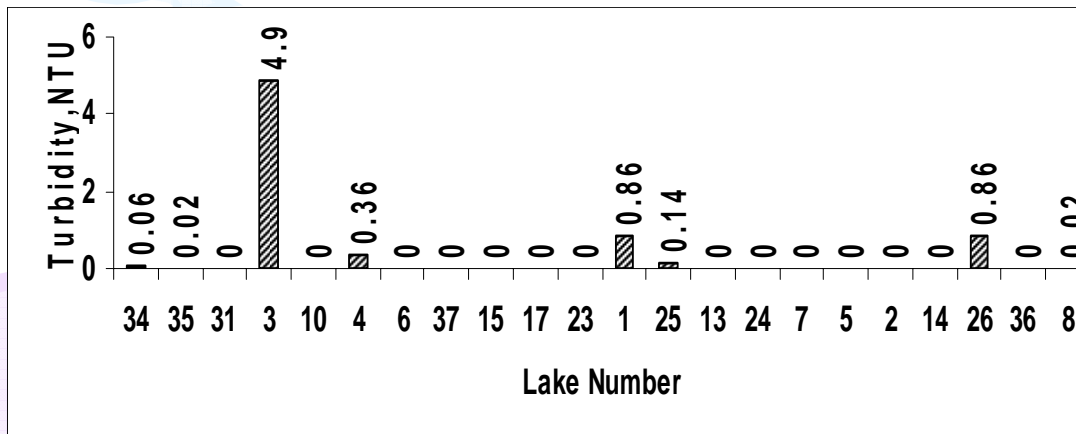
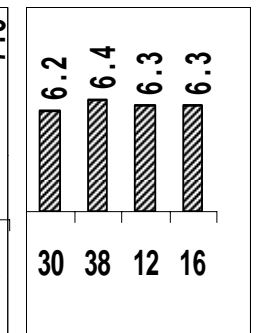
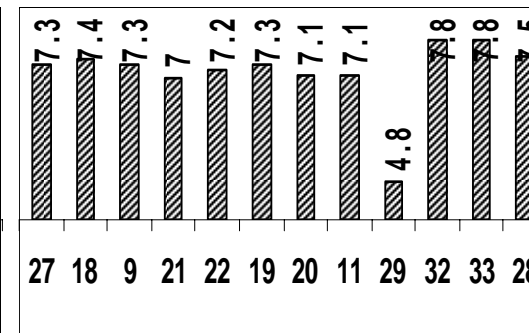
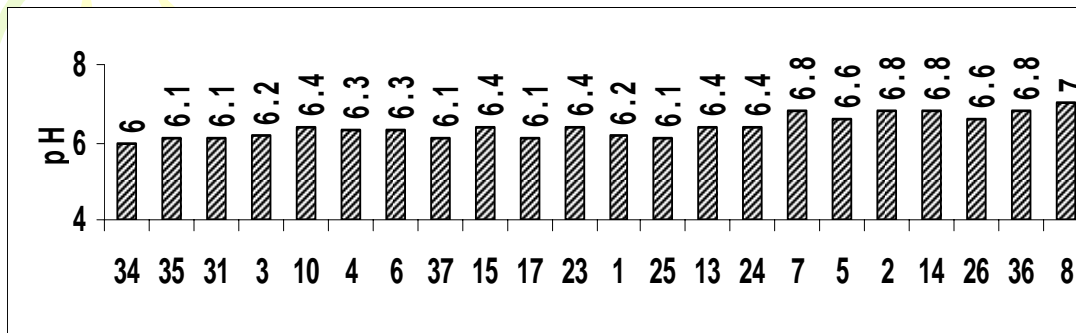
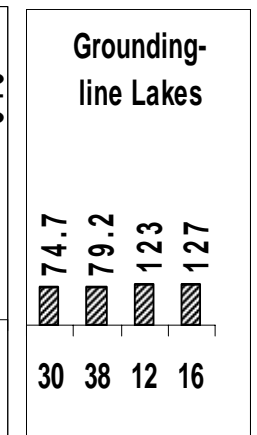
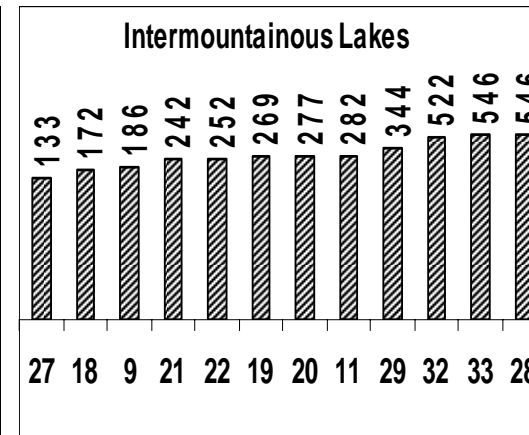
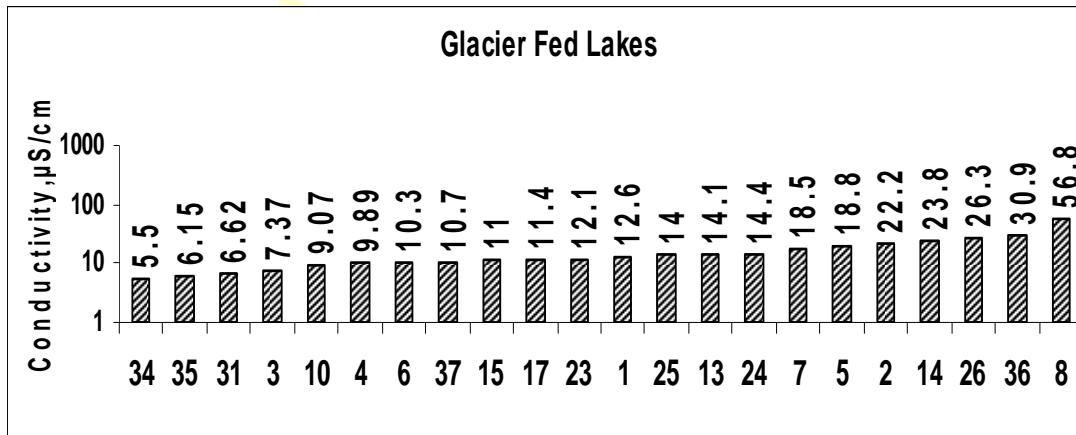
Sbrosovoye (13 m)



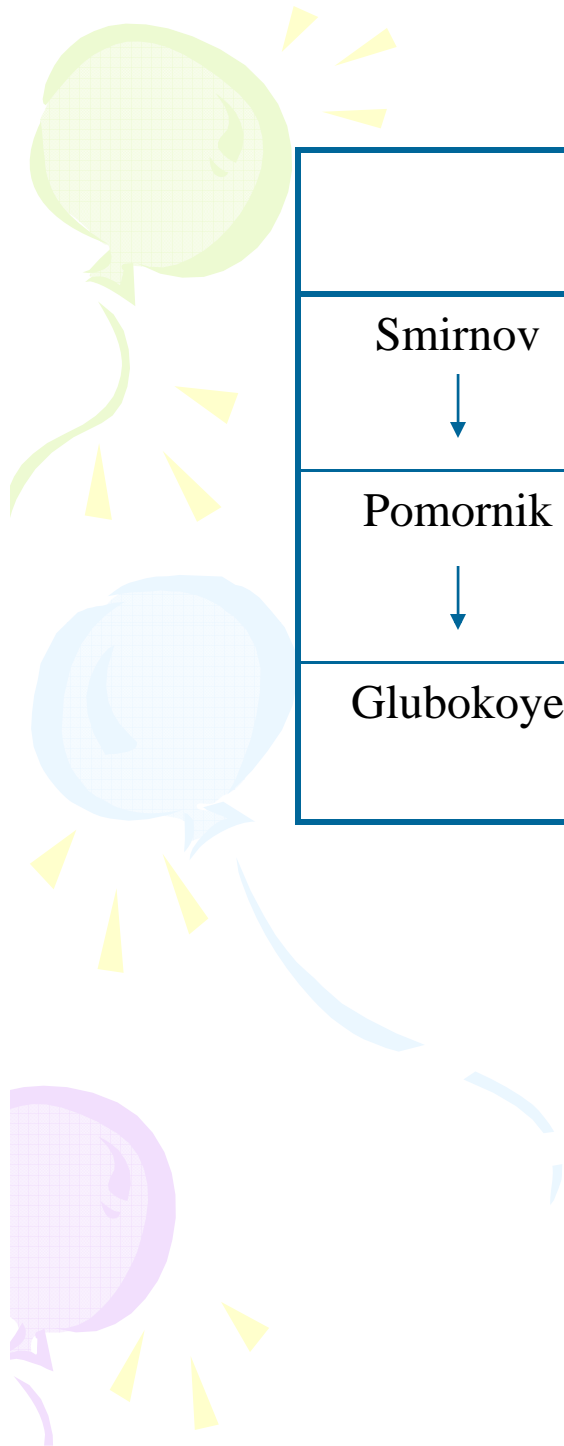
Smirnov (24 m)



Dlinnoye (5 m)



Some of the physical-chemical characteristics of Schirmacher Lakes



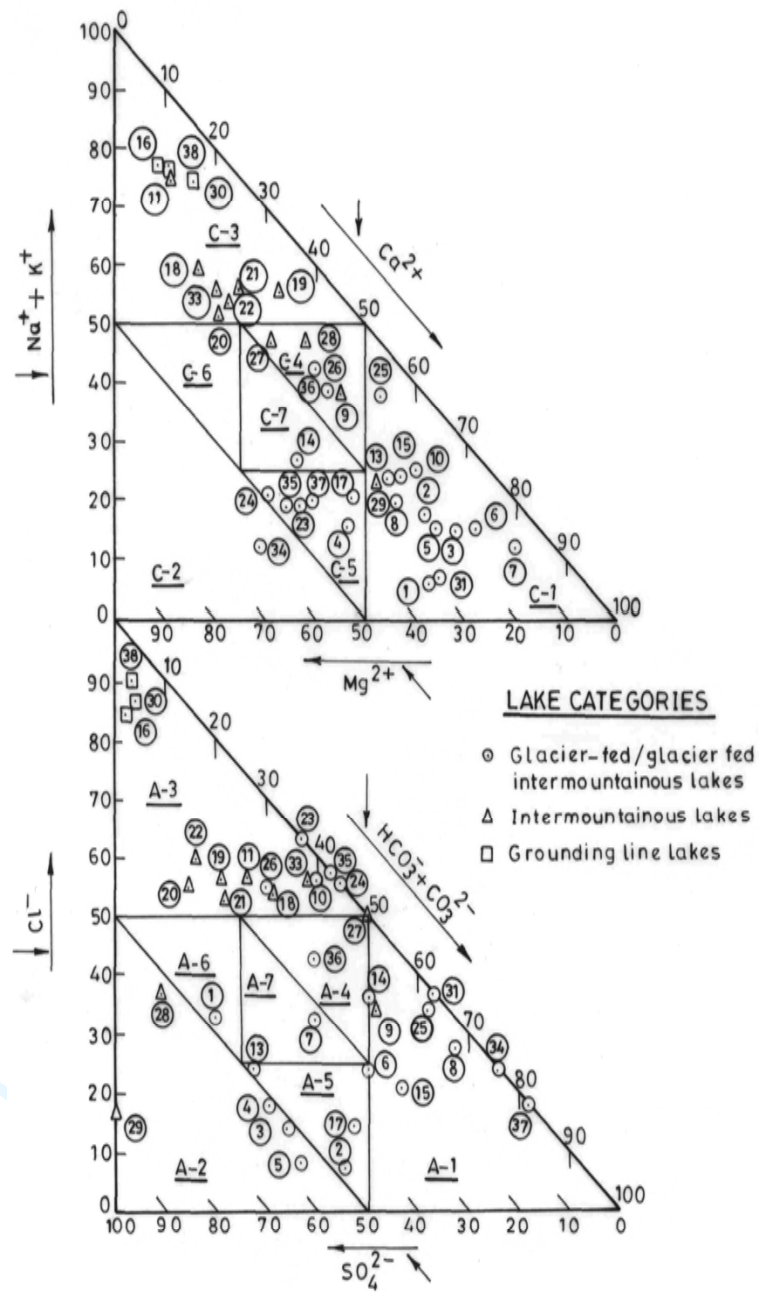
Glacier-fed		Glacier-fed Intermountainous	
Smirnov ↓	5.5 $\mu\text{S}/\text{cm}$ ↓	Zub ↓	14 $\mu\text{S}/\text{cm}$ ↓
Pomornik ↓	6.15 $\mu\text{S}/\text{cm}$ ↓	24 ↓	14.4 $\mu\text{S}/\text{cm}$ ↓
Glubokoye	10.7 $\mu\text{S}/\text{cm}$	26	26.3 $\mu\text{S}/\text{cm}$



Stone net, Sorted/ Stone/ Frost Polygon, Polygonal Ground/ Soil

Criteria for Classification of Lake Waters Based on Major Cations and Anions

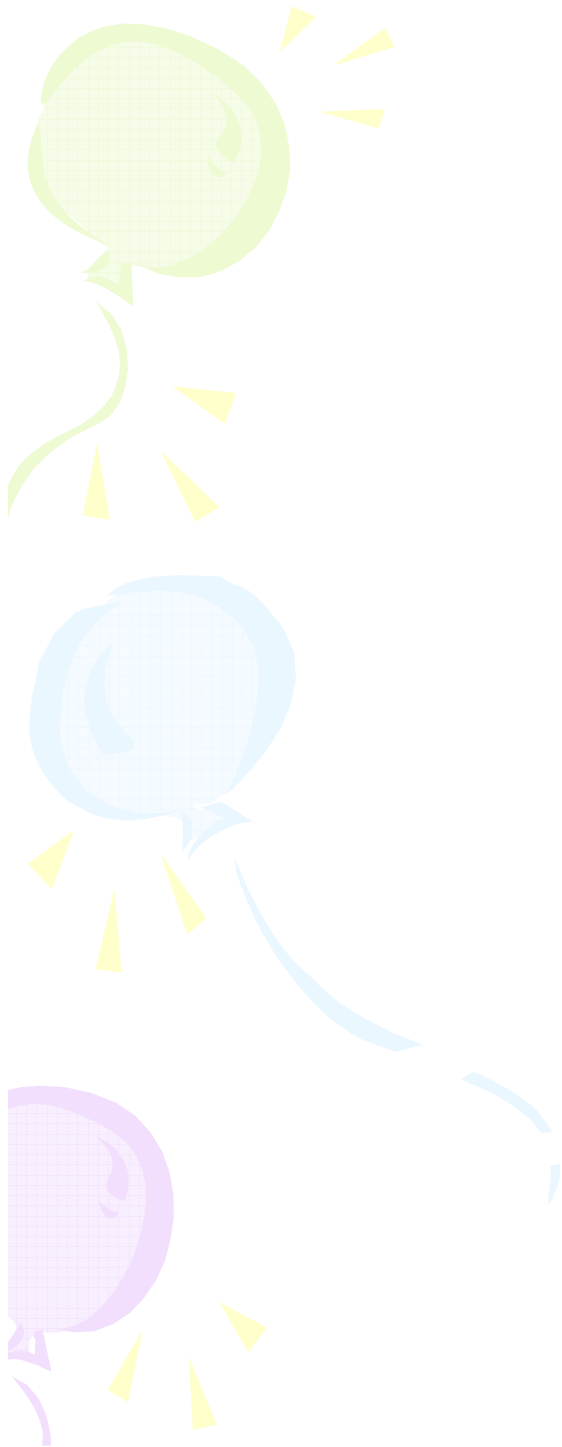
Sub-fields : Dominant Ions	Criteria
Cation triangle	
C1:Calcium	$\text{Ca}^{2+} > 50\%$
C2:Magnesium	$\text{Mg}^{2+} > 50\%$
C3:Sodium	$\text{Na}^+ > 50\%$
C4:Sodium-Calcium	$\text{Na}^+ \text{ \& \ } \text{Ca}^{2+}$ each 25-50%, $\text{Mg}^{2+} < 25\%$
C5:Calcium Magnesium	$\text{Ca}^{2+} \text{ \& \ } \text{Mg}^{2+}$ each 25-50%, $\text{Na}^+ < 25\%$
C6:Sodium Magnesium	$\text{Na}^+ \text{ \& \ } \text{Mg}^{2+}$ each 25-50%, $\text{Ca}^{2+} < 25\%$
C7:Calcium Magnesium Sodium	Ca^{2+} , Mg^{2+} & Na^+ each 25-50%
Anion Triangle	
A1:Bicarbonate	$\text{HCO}_3^- > 50\%$
A2:Sulphate	$\text{SO}_4^{2-} > 50\%$
A3:Chloride	$\text{Cl}^- > 50\%$
A4:Chloride-Bicarbonate	$\text{Cl}^- \text{ \& \ } \text{HCO}_3^-$ each 25-50%, $\text{SO}_4^{2-} < 25\%$
A5:Bicarbonate-Sulphate	$\text{HCO}_3^- \text{ \& \ } \text{SO}_4^{2-}$ each 25-50%, $\text{Cl}^- < 25\%$
A6:Chloride Sulphate	$\text{Cl}^- \text{ \& \ } \text{SO}_4^{2-}$ each 25-50%, $\text{HCO}_3^- < 25\%$
A7:Bicarbonate-sulphate chloride	HCO_3^- , SO_4^{2-} & Cl^- each 25-50%



Major Ion Chemistry of 38 Lake Waters from Schirmacher Oasis Represented in the form of Trilinear Cation-Anion

16 Water Types

Water Types
$\text{Ca}^{2+} - \text{HCO}_3^-$
$\text{Mg}^{2+} - \text{HCO}_3^-$
$\text{Ca}^{2+} - \text{SO}_4^{2-}$
$\text{Ca}^{2+} - \text{Mg}^{2+} - \text{Cl}^-$
$\text{Ca}^{2+} - \text{Cl}^-$
$\text{Ca}^{2+} - \text{Mg}^{2+} - \text{SO}_4^{2-}$
$\text{Ca}^{2+} - \text{Mg}^{2+} - \text{HCO}_3^- - \text{SO}_4^{2-}$
$\text{Ca}^{2+} - \text{SO}_4^{2-} - \text{Cl}^-$
$\text{Ca}^{2+} - \text{HCO}_3^- - \text{SO}_4^{2-}$
$\text{Ca}^{2+} - \text{HCO}_3^- - \text{SO}_4^{2-} - \text{Cl}^-$
$\text{Ca}^{2+} - \text{Mg}^{2+} - \text{Na}^+ - \text{HCO}_3^-$
$\text{Ca}^{2+} - \text{Na}^+ - \text{Cl}^-$
$\text{Ca}^{2+} - \text{Na}^+ - \text{HCO}_3^- - \text{Cl}^-$
$\text{Na}^+ - \text{Cl}^-$
$\text{Ca}^{2+} - \text{Na}^+ - \text{SO}_4^{2-}$
$\text{Na}^+ - \text{Ca}^{2+} - \text{HCO}_3^-$



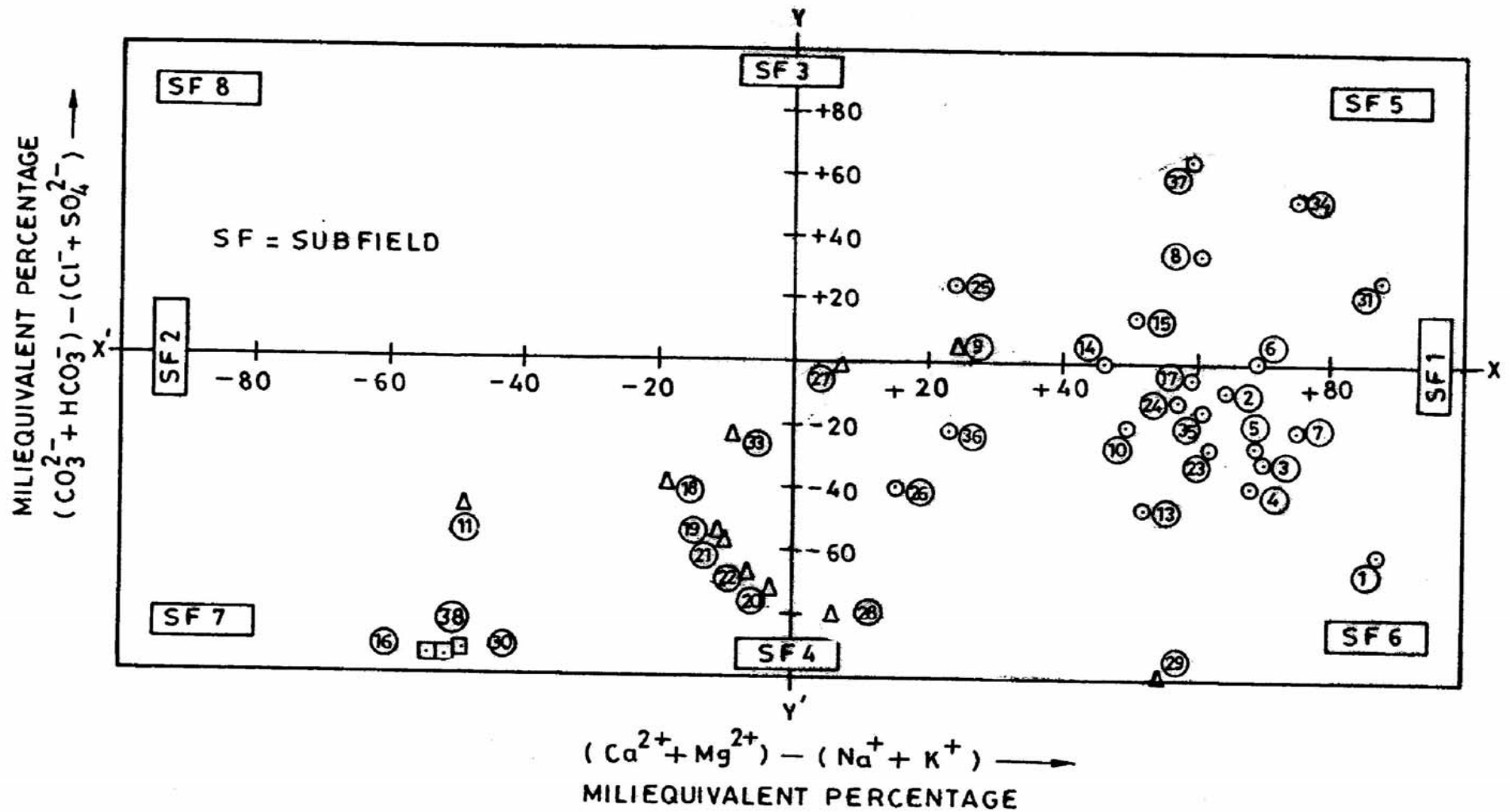
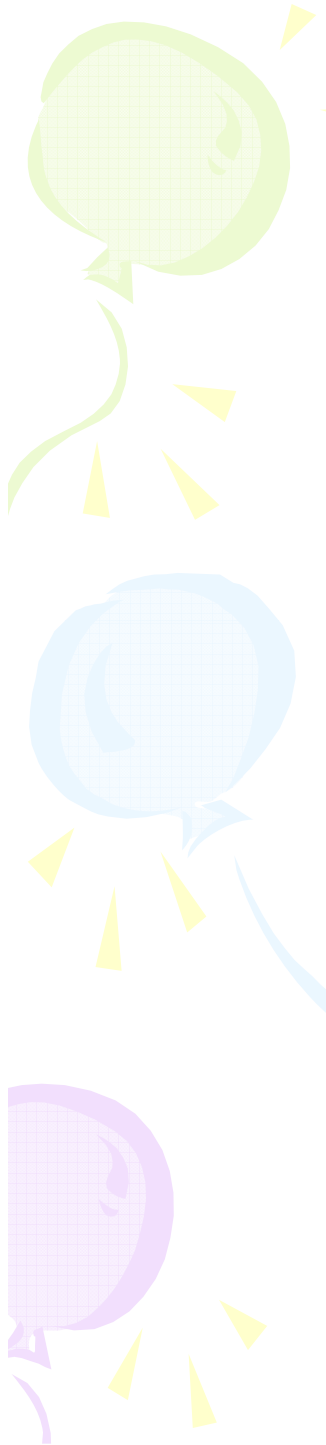


Fig. 4 Rectangular Ionic Field Describing the Overall Character of 38 Lake Waters from Schirmacher Oasis, Antarctica

Water Quality of Schirmacher Lakes : Summary of Results

Parameters	Schirmacher Lakes		
	Glacier-fed	Intermountainous	Grounding line
Conductivity, $\mu\text{S}/\text{cm}$ at 25°C	5.5 – 56.8	133-546	74.7-127
pH	6-7	4.8-7.8	6.2-6.4
Turbidity, NTU	ND-4.90	ND-0.76	ND
Total ions, meq/L	0.102-0.994	2.55-10.6	1.20-2.14
Dominant cation	Ca^{2+}	Na^{+}	Na^{+}
Dominant anion	HCO_3^{-}	Cl^{-}	Cl^{-}
Dominant water type	$\text{Ca}^{2+}\text{-HCO}_3^{-}$	$\text{Na}^{+}\text{-Cl}^{-}$	$\text{Na}^{+}\text{-Cl}^{-}$
Dominant water character	$\text{Ca}^{2+}+\text{Mg}^{2+} > \text{Na}^{+}+\text{K}^{+}$	$\text{Na}^{+}+\text{K}^{+} > \text{Ca}^{2+}+\text{Mg}^{2+}$	$\text{Na}^{+}+\text{K}^{+} > \text{Ca}^{2+}+\text{Mg}^{2+}$
	$\text{Cl}^{-}+\text{SO}_4^{2-} > \text{HCO}_3^{-}+\text{CO}_3^{2-}$	$\text{SO}_4^{2-}+\text{Cl}^{-} > \text{HCO}_3^{-}+\text{CO}_3^{2-}$	$\text{SO}_4^{2-}+\text{Cl}^{-} > \text{HCO}_3^{-}+\text{CO}_3^{2-}$
	$\text{NCH}^{*} > \text{CH}^{*}$	$\text{NCH} > \text{CH}$	$\text{NCH} > \text{CH}$

Every natural water has its own unique characteristics



*THANK
YOU*